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â€¢ To build structure, put in the bonds first, then distribute the rest of the valence electrons so that each atom gets an octet Example CO<sub>2</sub> need 3 atoms x 8 electrons = 24 e- have 4 e- + 2 x 6 e- = 16 e-----share 8 e-bonds 4 Chapter 5 Covalent Bonds and Molecular Structure pp 168-199

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between carbon atoms. a. Single Bonds. A single bond is a covalent bond formed by two carbon atoms sharing two electrons. Compounds that contain only single bonds between carbon atoms are called alkanes. b. Double Bonds. A double bond consists of two covalent bonds formed by two carbon atoms sharing four electrons as shown below.

## **LESSON ASSIGNMENT Paragraphs 3-1 through 3-18 and exercises.**

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